Al-Ir-Nb-Ni (Aluminum-Iridium-Niobium-Nickel)

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Recently, a schematic partial phase diagram was reported by [2005Hua] for this system with data obtained from alloy samples annealed between 1250 and 2000 °C.

Binary Systems

The Al-Ir phase diagram [2000Oka] depicts four intermediate phases: IrAl (*B*2, CsCl-type cubic), $IrAl_{2.7}$ (cubic), IrAl₃ (*D*0₁₈, Na₃As-type hexagonal), Ir_4Al_{13} (monoclinic), and Ir_2Al_9 (*D*8_d, Co₂Al₉-type monoclinic). For brief descriptions of the Al-Nb, Al-Ni, and Nb-Ni phase diagrams, see [2006Rag]. See [Massalski2] for the Ir-Nb phase diagram. Ir_3Nb has the $L1_2$ -type cubic structure. Ir and Ni form a continuous face-centered cubic (fcc) solid solution.

Ternary Systems

For the Al-Ir-Nb system, partial isothermal sections at 1100 °C were given by [1995Vil]. There appears to be no phase equilibrium data for the Al-Ir-Ni and Ir-Nb-Ni systems. An update of the Al-Nb-Ni system was given by [2006Rag].

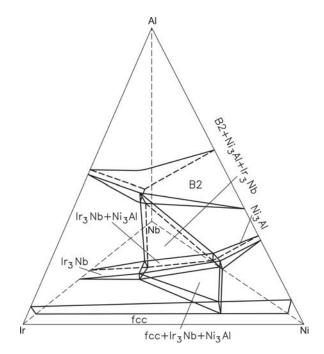


Fig. 1 Al-Ir-Nb-Ni schematic partial phase equilibria at \geq 1250 °C [2005Hua]

Quaternary Phase Equilibria

In a series of papers [1999Yu, 2000Yu, 2002Hua, 2003Hua, 2005Hua], Huang et al. investigated the phase equilibria of this quaternary system. [2005Hua] arc melted 12 Ni-rich alloys and six Ir-rich alloys and annealed them up to 168 h between 1250 and 2000 °C. The phase equilibria were studied by scanning and transmission electron microscopy, x-ray diffraction, differential thermal analysis, and electron probe microanalysis. A schematic (partial) phase diagram applicable for $\geq 1250^{\circ}$ C was constructed by [2005Hua] within the composition tetrahedron, Fig. 1. The continuous fcc solid solution between Ni and Ir dissolve some Al and Nb. The B2-type phases NiAl and IrAl also form a continuous solution. Ir₃Nb and Ni₃Al both with the L1₂-type structure show only limited solubility in each other, presumably due to the large difference in the lattice parameters (a = 0.38915 nm for Ir₃Nb and 0.35720 nm for Ni₃Al). Figure 1 shows two three-phase regions $Ir_3Nb +$ $Ni_3Al + fcc$ and $Ir_3Nb + Ni_3Al + B2$. Another related system Al-Ir-Nb-Pt investigated by [2005Hua] showed a continuous L_{1_2} -type solid solution between Ir₃Nb and Pt₃Al (a =0.38775 nm for Pt₃Al).

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